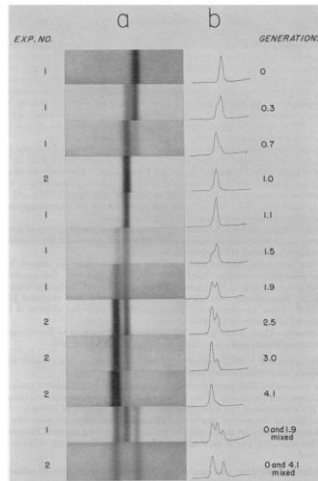


Semi-conservative replication of DNA - The experiment of Meselson & Stahl



"Timing, hard work
& Serendipity"

1958

2

INTRODUCTION - BIOCHEMISTRY THE PRECURSORS

3

The Structure of the Hereditary Material

An account of the investigations which have led to the formulation of an understandable structure for DNA. The chemical reactions of this material within the nucleus govern the process of reproduction

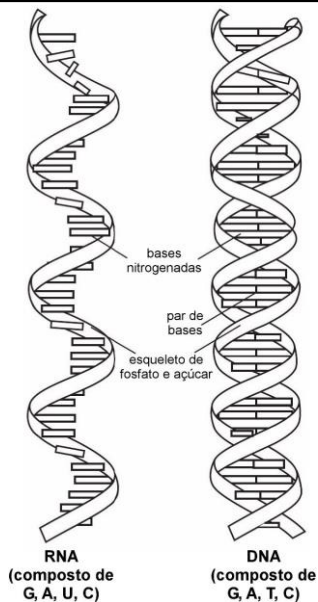
by F. H. C. Crick

If we knew the monomers from which nature makes DNA, RNA and protein, we might be able to carry out very spectacular experiments in the test tube. Be that as it may, we now have for the first time a well-defined model for DNA and for a possible replication process, and this in itself should make it easier to devise crucial experiments.

© 1954 SCIENTIFIC AMERICAN, INC

4

Deoxyribonucleic acid (DNA) and ribonucleic acid (RNA)



- 1909 the word gene is introduced by Johannsen (1857-1927)
- 1941 Beadle (1903-1989) & Tatum (1909-1975) -> 1 gene:1enzyme.
-> Nobel prize 1958
- 1944 Oswald Avery (1877-1955) – DNA is the genetic material -> Never received a Nobel prize.
- 1952 Hershey (1908-1997) & Chase (1927-2003) – Hershey Nobel 1969
- 1953 Watson (1928-) & Crick (1916-2004) – model of DNA double helix -> Nobel prize 1962
- 1958 Crick – “Central Dogma”

5

The central dogma according to Francis Crick

"... once 'information' has passed into protein *it cannot get out again.*"



6

The central dogma: an oversimplification?

NATURE VOL. 227 AUGUST 8 1970

"The central dogma, enunciated by Crick in 1958 and the keystone of molecular biology ever since, is likely to prove a considerable over-simplification."

561

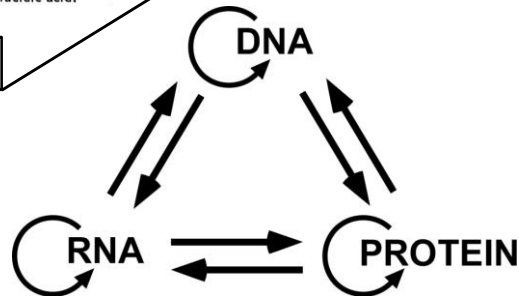
Central Dogma of Molecular Biology

by
FRANCIS CRICK
MRC Laboratory of Molecular Biology,
Hills Road,
Cambridge CB2 2QH

The central dogma of molecular biology deals with the detailed residue-by-residue transfer of sequential information. It states that such information cannot be transferred from protein to either protein or nucleic acid.

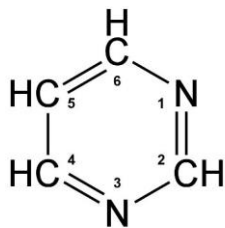
"The central dogma, enunciated by Crick in 1958 and the keystone of molecular biology ever since, is likely to prove a considerable over-simplification."

This quotation is taken from the beginning of an unsigned article¹ headed "Central dogma reversed", recounting the very important work of Dr Howard Temin² and others³ showing that an RNA tumour virus can use viral RNA as a template for DNA synthesis.

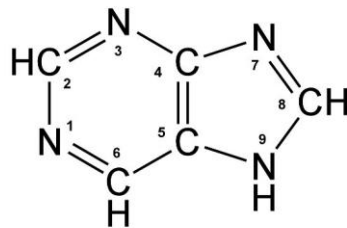


7

Bases that make up the nucleic acids



PYRIMIDINE

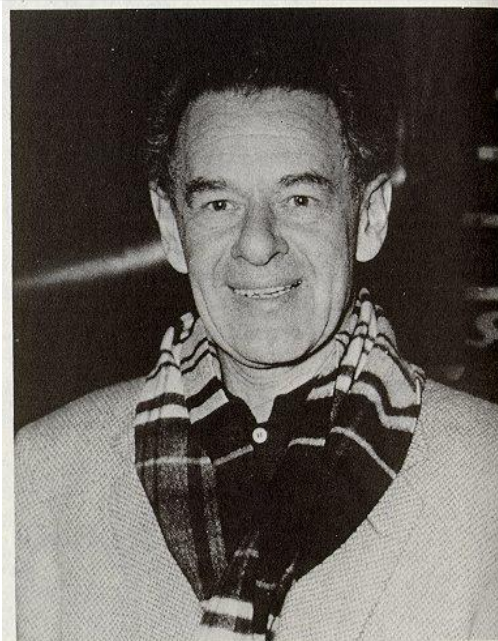


PURINE

Numbering of atoms in the nitrogenous bases

8

- 1953 -


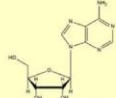
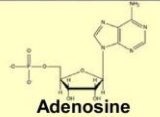

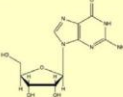
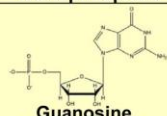

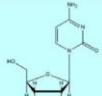
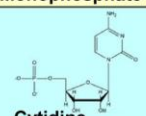


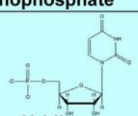


**Fritz Lipmann
(1899-1986)**

NPW 1953

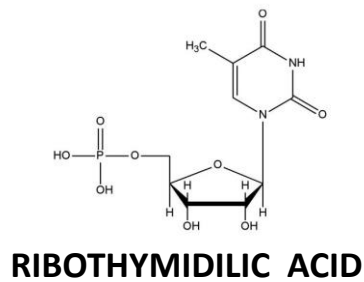
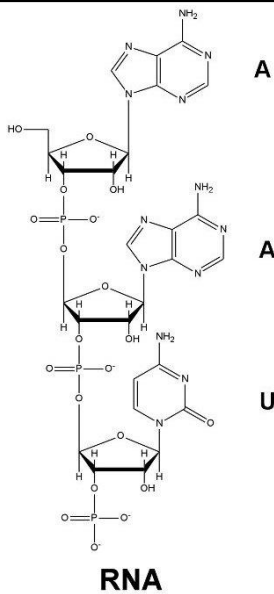
9

RNA Mononucleotide components

	BASE	NUCLEOSIDE	NUCLEOTIDE
PURINES	 Adenine	 Adenosine	 Adenosine Monophosphate
	 Guanine	 Guanosine	 Guanosine Monophosphate
PYRIMIDINES	 Cytosine	 Cytidine	 Cytidine Monophosphate
	 Uracil	 Uridine	 Uridine Monophosphate

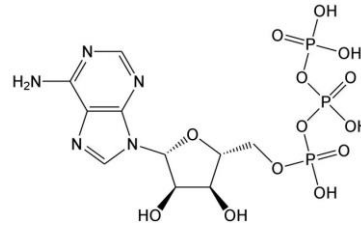
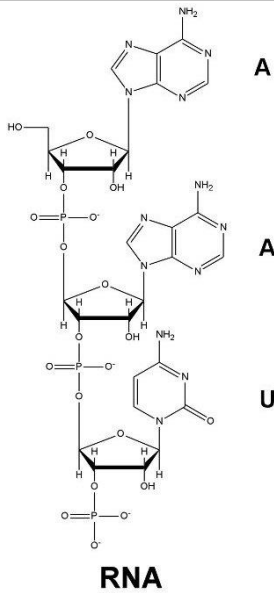
12

RNA and some nucleotides



13

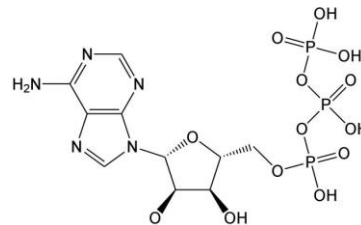
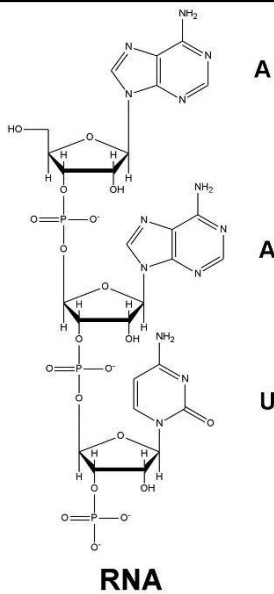
RNA and some nucleotides



ATP (adenosine triphosphate)

14

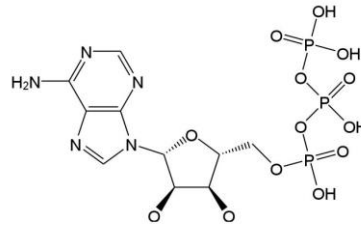
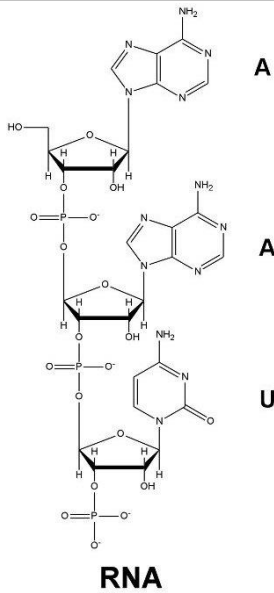
RNA and some nucleotides



dATP (2'-deoxyadenosine triphosphate)

15

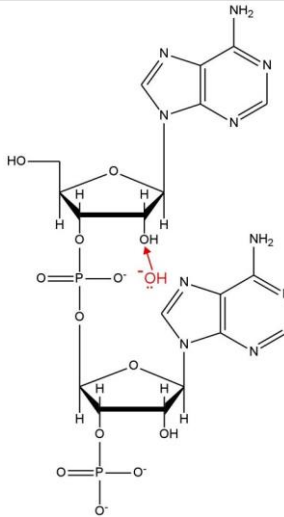
RNA and some nucleotides



ddATP (dideoxyadenosine triphosphate)

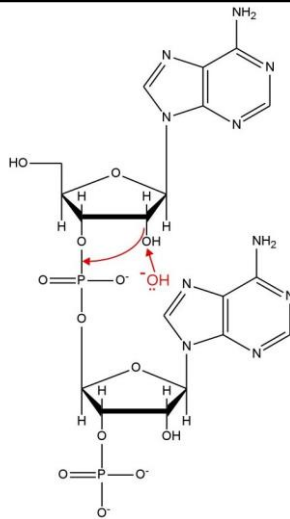
16

Polyribonucleotides are degraded by alkali



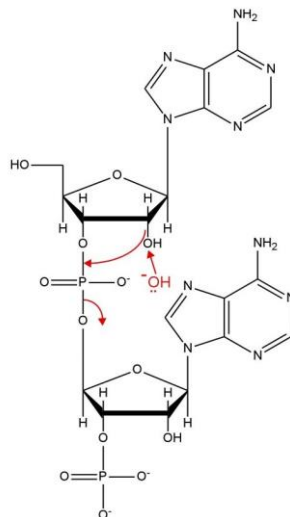
17

Polyribonucleotides are degraded by alkali



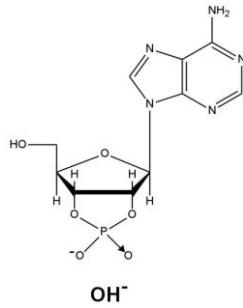
18

Polyribonucleotides are degraded by alkali



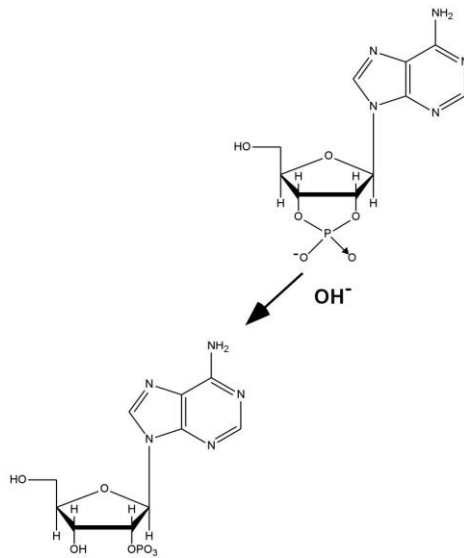
19

Polyribonucleotides are degraded by alkali



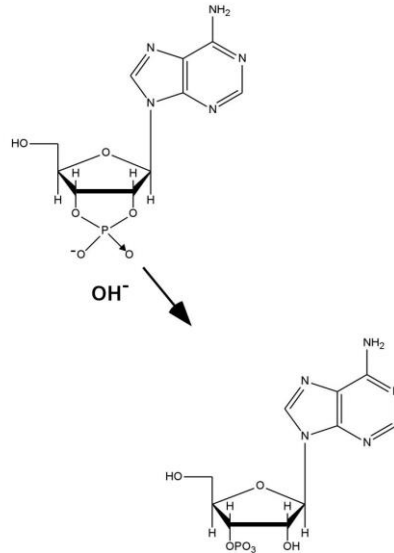
20

Polyribonucleotides are degraded by alkali



21

Polyribonucleotides are degraded by alkali



22

EXPERIENTIA

Vol. VI - Fasc. 6

Pag. 201-240

15. VI. 1950

Chemical Specificity of Nucleic Acids and Mechanism of their Enzymatic Degradation¹

By ERWIN CHARGAFF², New York, N.Y.

Erwin Chargaff (1905-2002)

$$A + G = T + C$$

or

$$\frac{A + G}{T + C} = 1$$

$$A = T$$

$$G = C$$

thus,

$$A + C = G + T$$

23

Chargaff 1950

*Table IV*⁵
Composition of two microbial deoxyribonucleic acids.

Constituent	Yeast		Avian tubercle bacilli
	Prep. 1	Prep. 2	
Adenine	0.24	0.30	0.12
Guanine	0.14	0.18	0.28
Cytosine	0.13	0.15	0.26
Thymine	0.25	0.29	0.11
Recovery	0.76	0.92	0.77

*Table III*²
Composition of deoxyribose nucleic acid of man (in moles of
nitrogenous constituent per mole of P).

Constituent	Sperm		Thymus	Liver	
	Prep. 1	Prep. 2		Normal	Carcinoma
Adenine . . .	0.29	0.27	0.28	0.27	0.27
Guanine . . .	0.18	0.17	0.19	0.19	0.18
Cytosine . . .	0.18	0.18	0.16		0.15
Thymine . . .	0.31	0.30	0.28		0.27
Recovery . . .	0.96	0.92	0.91		0.87

Chargaff E (1950) *Experientia* 6:201-240

24

most students no longer study nature; they test models.

Erwin Chargaff

One of the obnoxious dogmas to which it has given rise - the so-called Central Dogma: DNA makes RNA; RNA makes proteins - is no longer valid. (I had never accepted it, as shown in lectures I gave in 1957 in Moscow and in 1958 in Vienna)
But the fact that dogmas could be handed down from the mountains shows that science had changed disastrously.

Erwin Chargaff

25

I have a vivid recollection of a 1956 discussion with two distinguished biochemists, X and Y, about the significance of the Watson-Crick structure of DNA. I was not making much of an impression. Finally I asked X, ..."Do you believe that the Watson-Crick structure is essentially correct?" The amazing answer: "Yes, I think it is correct, but I don't think it has anything to do with replication."

George Beadle, 1966

26

How DNA duplicates, according to Max Delbrück

ON THE REPLICATION OF DESOXYRIBONUCLEIC ACID (DNA)

BY M. DELBRÜCK

KERCKHOFF LABORATORIES OF BIOLOGY, CALIFORNIA INSTITUTE OF TECHNOLOGY

Communicated May 18, 1954

The discoveries of Hershey and Chase¹ concerning the role of DNA in transmitting genetic information in phage and of Watson and Crick² concerning the structure of DNA have brought the problem of the replication of DNA into focus. The structure proposed by Watson and Crick consists of two polynucleotide chains wound helically around a common axis, tied together by hydrogen bonds between the

Delbrück M (1954) Proc. Natl. Acad. Sci. USA 40:783-787

27

How DNA duplicates, according to Max Delbrück

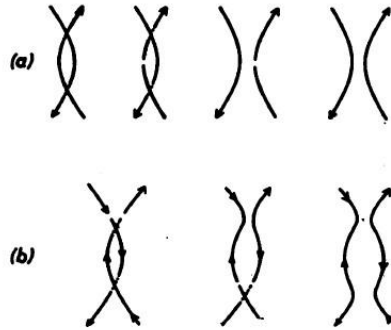


FIG. 2.—Two methods of resolving an interlock between two chains: *a*, By breaking one chain and slipping the other through the gap. *b*, By breaking both chains at each overlap and rejoining them criss-cross. *Two pairs of breaks* are needed to resolve one interlock.

Delbrück M (1954) Proc. Natl. Acad. Sci. USA 40:783-787

28

How DNA duplicates, according to Max Delbrück

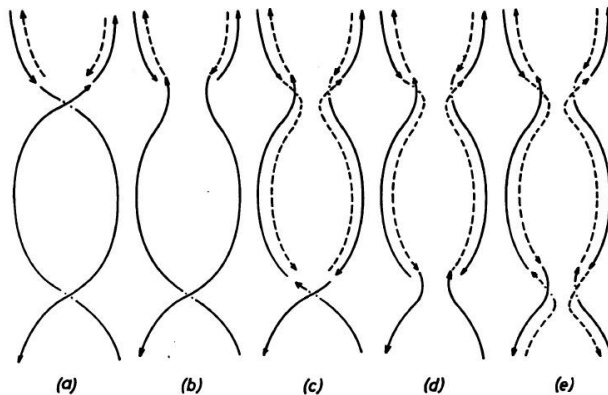


FIG. 3.—Resolution of an interlock in a replicating duplex by breaking both old chains at each half-turn of the helix and rejoining the lower terminals of the breaks to the open ends of equal polarity of the new chains. Lateral view. *a*, Location of first pair of breaks. *b*, Rejoining of lower terminals of breaks. *c*, Location of second pair of breaks. *d*, Rejoining of lower terminals.

Parental chains are represented by solid lines; new chains, by dashed lines. At the overlaps the lower chains are dotted.

Delbrück M (1954) Proc. Natl. Acad. Sci. USA 40:783-787

29

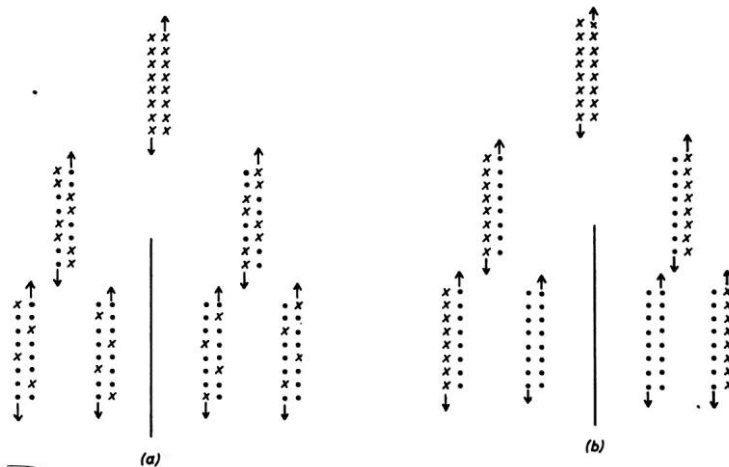
How DNA duplicates, according to Max Delbrück

"If a labeled duplex replicates repeatedly at the expense of an unlabeled pool, then according to this model, the label will be statistically equally distributed to the daughter-duplexes at each successive replication."

Delbrück M (1954) *Proc. Natl. Acad. Sci. USA* 40:783-787

30

How DNA duplicates, according to Max Delbrück



"Distribution of labeled parental chains to daughter-duplexes, with breaks and rejoins as postulated in the theory here presented"

Delbrück M (1954) *Proc. Natl. Acad. Sci. USA* 40:783-787

31

Precursor (7 months before M&S !!!)

THE ORGANIZATION AND DUPLICATION OF CHROMOSOMES AS REVEALED BY AUTORADIOGRAPHIC STUDIES USING TRITIUM-LABELED THYMIDINE

BY J. HERBERT TAYLOR,* PHILIP S. WOODS, AND WALTER L. HUGHES

DEPARTMENT OF BOTANY, COLUMBIA UNIVERSITY; BIOLOGY DEPARTMENT AND
MEDICAL DEPARTMENT, BROOKHAVEN NATIONAL LABORATORY

Communicated by Franz Schrader, October 26, 1956

32

Precursor (7 months before M&S !!!)

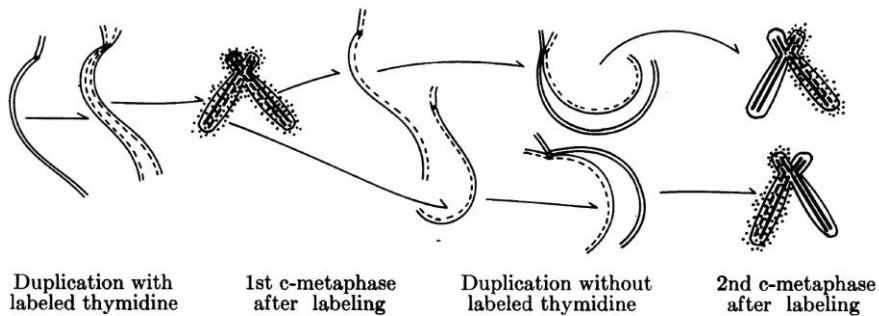


FIG. 3.—Diagrammatic representation of proposed organization and mode of replication which would produce the result seen in the autoradiographs. The two units necessary to explain the results are shown, although these were not resolved by microscopic examination. Solid lines represent nonlabeled units, while those in dashed lines are labeled. The dots represent grains in the autoradiographs.

It is immediately apparent that this pattern of replication is analogous to the replicating scheme proposed for DNA by Watson and Crick.⁶ We cannot be sure,

33

Precursor (7 months before M&S !!!)

Summary.—Tritium-labeled thymidine was prepared and used for labeling chromosomes during their duplication. Analysis of autoradiographs showed that both daughter chromosomes resulting from duplication in the presence of labeled thymidine appeared equally and uniformly labeled. After an ensuing duplication in the absence of the labeled DNA precursor, the label appeared in only one of each two chromatids (daughter chromosomes). These findings indicate that DNA is synthesized as a unit which extends throughout the length of the chromosome. The units remain intact through succeeding replications and nuclear divisions, except for occasional chromatid exchanges. Each chromosome is composed of two such units, probably complementary to each other. After each replication the four resulting units separate, so that each daughter chromosome always contains an “original” and a “new” unit. To explain the results, a model with two complementary units and a scheme of replication analogous to the Watson-Crick model of DNA is proposed.

34

THE CHARACTERS AND THE PROBLEM

35

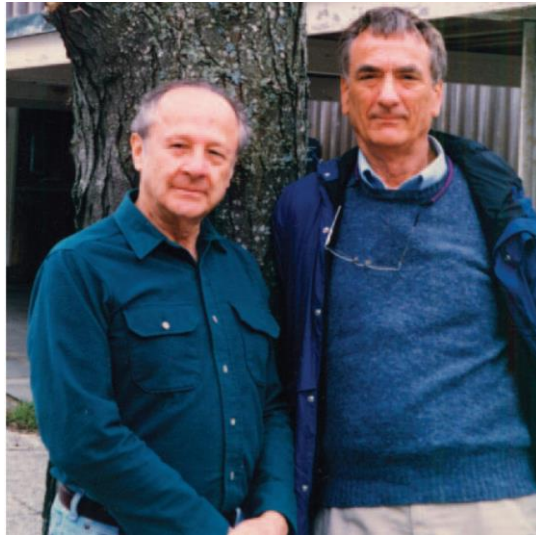
MATTHEW MESELSON**Born:** 24 de maio de 1930, em Denver, CO**PhD Advisor (1957):** Linus Pauling**PhD at:** California Institute of Technology (CalTech)**PhD thesis on:** (1) Ultracentrifugation; (2)
Crystallography**FRANKLIN W. STAHL****Born:** 8 de outubro de 1929 em Boston, MA**PhD Advisor (1956):** August H. Doermann**PhD at:** University of Rochester**PhD thesis on:** Genetics of Bacteriophage T4

36

<https://ibiology.org><https://youtu.be/V2evjmkur7k?si=ISj7865Wvc9E9RSE>Meselson & Stahl interview -> <https://youtu.be/7-tnuAqEp9g?si=A1p9If3TODSBwAOt>Semi-conservative replication cartoon -> <https://youtu.be/8jPK3S9S8rg?si=VWUB5R96kUgTLLgv>DNA replication – 3D animation -> <https://youtu.be/UpNZws-G8HQ?si=UZlnpXUqbdOYdBml>Stephen Bell – Replication fork -> <https://youtu.be/TweBOe3DIfY?si=hMVd-DCUMEGR3OO7>

37

Matt Meselon and Frank Stahl in 1996 under the tree in Woods Hole where they decided the famous experiment 42 years earlier



Hanawalt, PNAS **101**:17889-17894, 2004

38

THEORETICAL AND METHODOLOGICAL BASES

39

The problem of macromolecule transport

DIFFUSION is the transport of matter in a mixture caused by concentration gradients

STOKES-EINSTEIN-SUTHERLAND EQUATION (1905)

$$D = \frac{k_B T}{6\pi \eta r}$$

D = diffusion coefficient

k_B = Boltzman constant

T = absolute temperature

η = Dynamic viscosity

r = radius of the spherical particle

Constante dos Gases (R)
Número de Avogadro (N)

SEDIMENTATION is the transport of matter in a mixture due to a external field, in particular due to gravity force or to centrifugal force.

40

THEODOR SVEDBERG [1884-1971]



- When 15 years old constructs a Marconi transmitter
- When 16 years old begins the studies at the University of Uppsala
- First paper published when he was 17 years old
- PhD at age 19

Chemistry Nobel Prize in 1926, age 42

Main interest: colloidal solutions

Helped the foundation of LKB that later fused with Pharmacia and today is part of GE Lifesciences

Married four times, had 12 children and published 229 papers (the last one when aged 81) and 13 books!

41

Dec., 1924

THE ULTRA-CENTRIFUGE

2677

[CONTRIBUTION FROM THE LABORATORY OF PHYSICAL CHEMISTRY OF THE UNIVERSITY OF
UPSALA]

THE ULTRA-CENTRIFUGE, A NEW INSTRUMENT FOR THE DETERMINATION OF SIZE AND DISTRIBUTION OF SIZE OF PARTICLE IN AMICROSCOPIC COLLOIDS

BY THE SVEDBERG AND HERMAN RINDE

RECEIVED JULY 17, 1924

PUBLISHED DECEMBER 13, 1924

"The new centrifuge constructed by us allows the determination of particles that cannot be made visible in the ultramicroscope. In analogy with the naming of the [ultra-microscope](#) and [ultra-filtration](#) apparatus **we propose the name ultra-centrifuge** for this apparatus."

42

Svedberg's "ultracentrifuge" attained a maximum of 5000 g

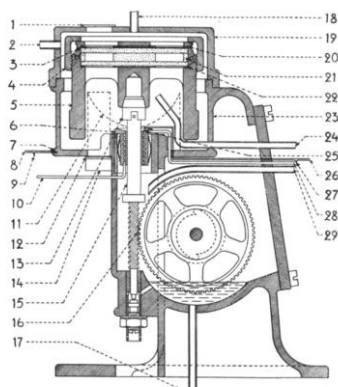


Fig. 1.

1, Upper window; 2, Hydrogen outlet; 3, Ebonite plate; 4, Lid of rotor; 5, Rotor; 6, Water-cooled spring bearing; 7-9, Thermocouple; 10, Water outlet; 11, Copper screen; 12, Lower window; 13, Carrying cone; 14, Reflecting prism; 15, Shaft of rotor; 16, Toothed wheel; 17, Oil outlet; 18, Hydrogen inlet; 19, Lid; 20, Rubber plate; 21, Cell; 22, Rubber plate; 23, Casing of centrifuge; 24, Hydrogen inlet; 25-27, Thermocouple; 28, Water inlet; 29, Oil inlet.

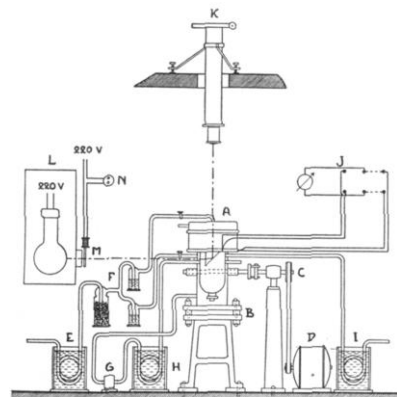


Fig. 2.

A, Centrifuge; B, Stand; C, Pulley; D, Motor; E, Cooling spiral for hydrogen current; F, Valves; G, Oil circulation pump; H, Cooling spiral for oil current; I, Cooling spiral for water current; J, Galvanometer and key; K, Camera; L, Lamp; M, Shutter; N, Switch for shutter.

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430

THE SVEDBERG AND ROBIN FÅHRAEUS

Vol. 48

[CONTRIBUTION FROM THE LABORATORY OF PHYSICAL CHEMISTRY OF THE UNIVERSITY OF
UPSALA]

A NEW METHOD FOR THE DETERMINATION OF THE MOLECULAR WEIGHT OF THE PROTEINS

BY THE SVEDBERG AND ROBIN FÅHRAEUS

RECEIVED AUGUST 4, 1925

PUBLISHED FEBRUARY 5, 1926

44

The Svedberg equation

$$M = \frac{RT}{D(1 - \bar{v}\rho)} \frac{dr/dt}{\omega^2 r}$$

M = molecular mass

D = diffusion coefficient

R = gas constant (8,3143x10⁷ ergs/grau mol)

T = absolute temperature (in K)

r = distance from the rotation center to the center of the band

ω = angular velocity

ρ = solvent density

\bar{v} = protein partial specific volume

45

The Svedberg equation

$$\frac{dr/dt}{\omega^2 r} = s = \text{sedimentation constant}$$

$$M = \frac{RTs}{D(1 - \bar{v}\rho)}$$

s is measured in reciprocal seconds

s basic unit 10^{-13} s, is called “one Svedberg (S)”

46



Analytical Ultracentrifuge Beckman/Spenco – (Scientific American, 1951)

A RESEARCH TOOL for the centrifugal purification and molecular-weight determination of such macromolecules as proteins, viruses, polymers, enzymes, hormones, and chemical precipitates; the Spenco Model E Ultracentrifuge combines all operations in a single cabinet.

OPERATING FEATURES include a 70,000-rpm quiet electric drive designed for years of service and arranged for easy removal and exchange; a selection of analytical and preparative rotors which are interchangeable and provide forces up to 260,000 times gravity operating in a heavily-armored vacuum rotor chamber; an optical system of the Philpot-Svensson refractive-index type with viewing screen, automatic camera, and high-intensity light source for analytical work; a

refrigerating system for keeping rotors at controlled temperatures; and vacuum pumps which provide atmospheres down to 10^{-4} mm of mercury.

CONTROLS AND INSTRUMENTS are included to maintain average drive-speed accuracy automatically within 0.1 per cent of setting; to set operation of automatic camera both for exposure and photo-interval; to read rotor temperatures; to select one of three graded braking rates; to read motor voltage and current; to read vacuum; and to actuate the camera manually. Interlocks are provided to insure safe operation by individuals without extensive training.

Send for complete details.

SPINCO = Specialized Instruments Corp.

Spenco division

BECKMAN INSTRUMENTS, INC.
BELMONT 5, CALIFORNIA



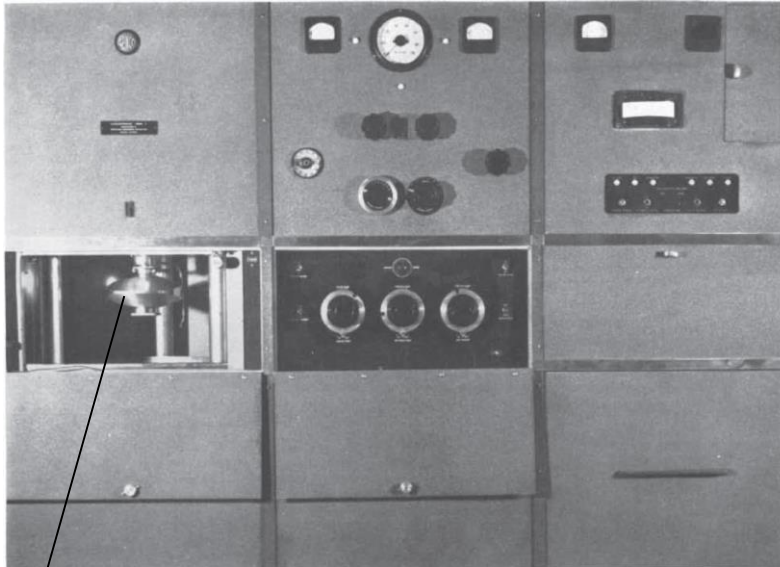
24A

For further information, circle number 24 & on Reader's Service Card, page 51A.

ANALYTICAL CHEMISTRY

47

Panel of an analytical ultracentrifuge in the 1950s



ROTOR

Gray, *Scient. Amer.* **184**(6):42-51, 1951

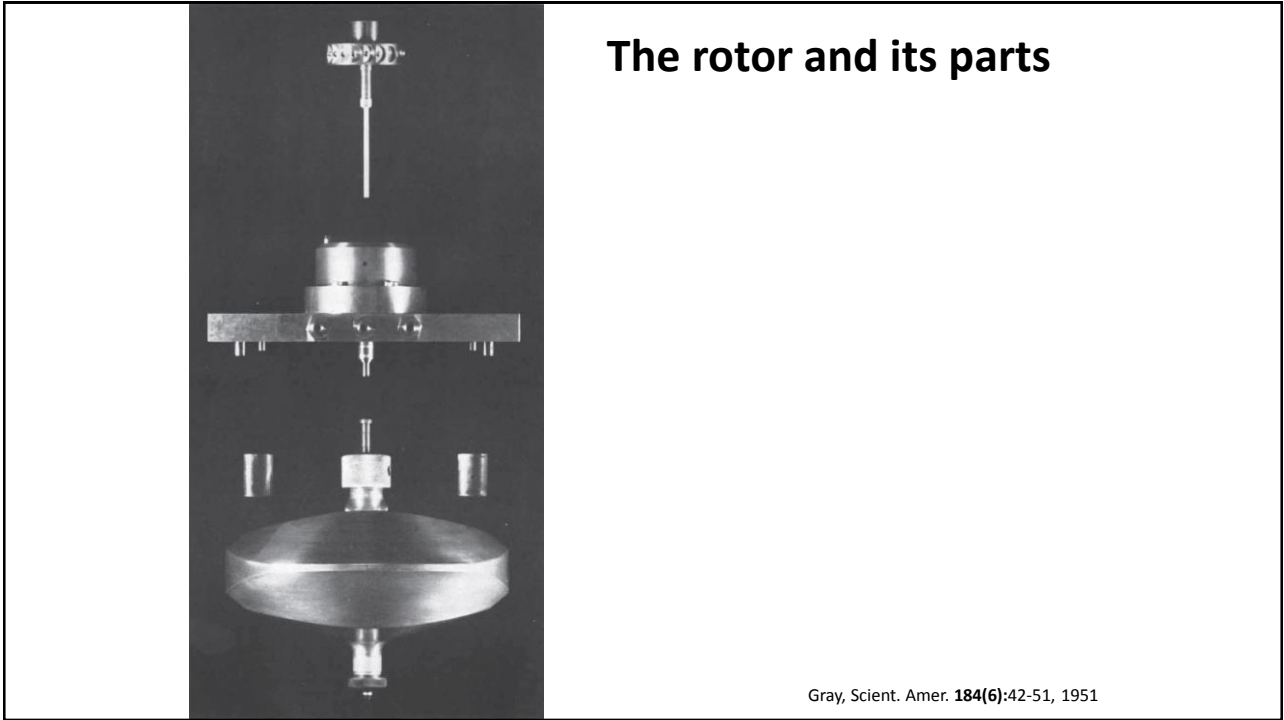
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SPINCO Model E Analytical Ultracentrifuge

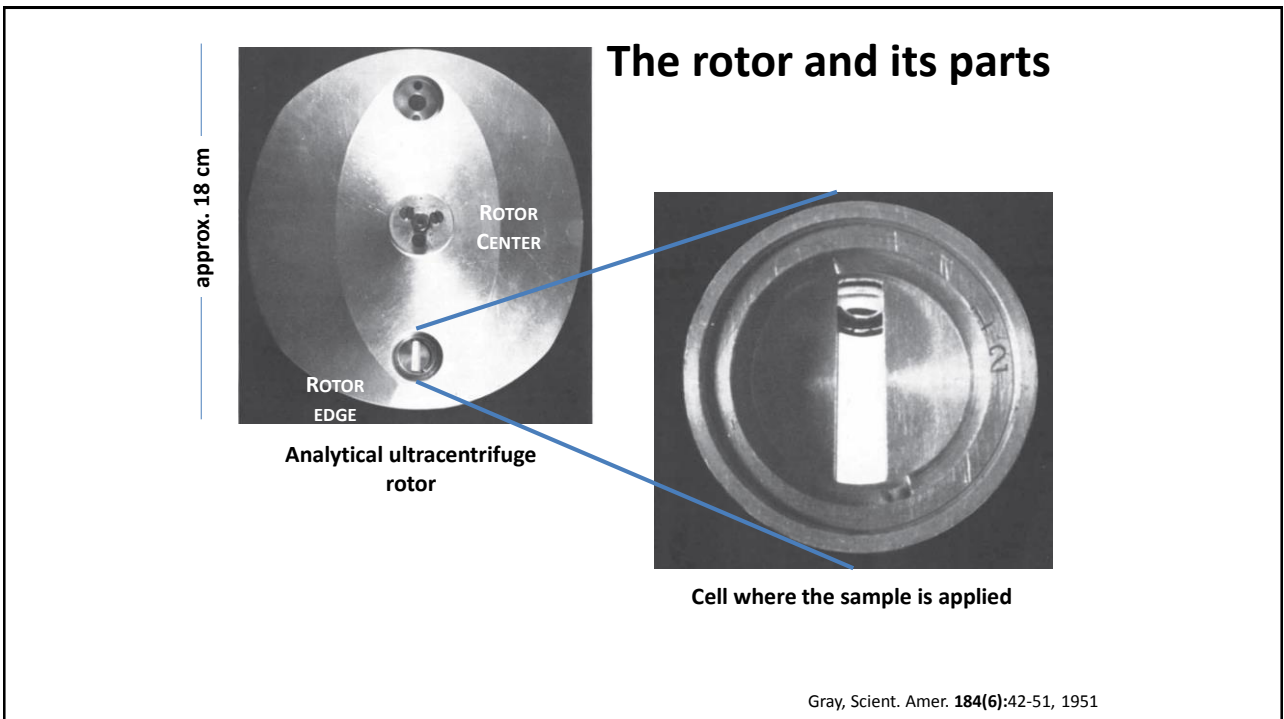


Planken KL (2008) – PhD Thesis - ISBN 978-90-393-4798-0

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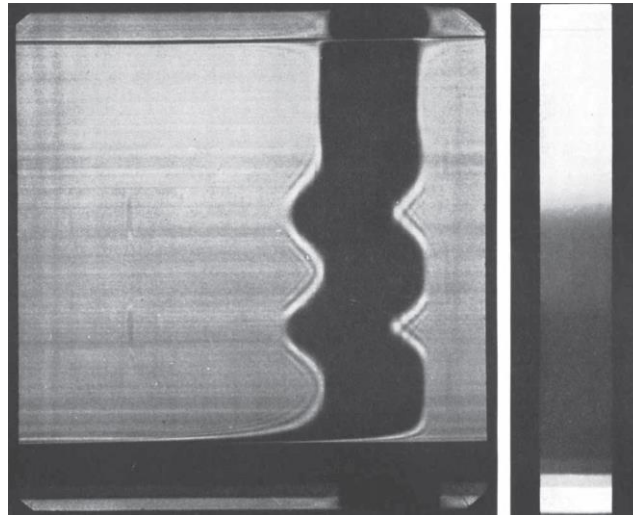
50



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Observing the behavior of two proteins

ROTOR
CENTER
TOP



BOTTOM

ROTOR
EDGE

Photography of the Refraction pattern

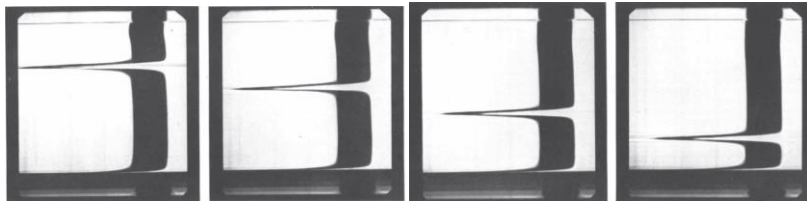
Absorption

Gray, *Scient. Amer.* **184**(6):42-51, 1951

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Ultracentrifugation of *Limulus polyphemus* hemocyanin

ROTOR
CENTRE



ROTOR
EDGE

1:00 h

1:30 h

2:00 h

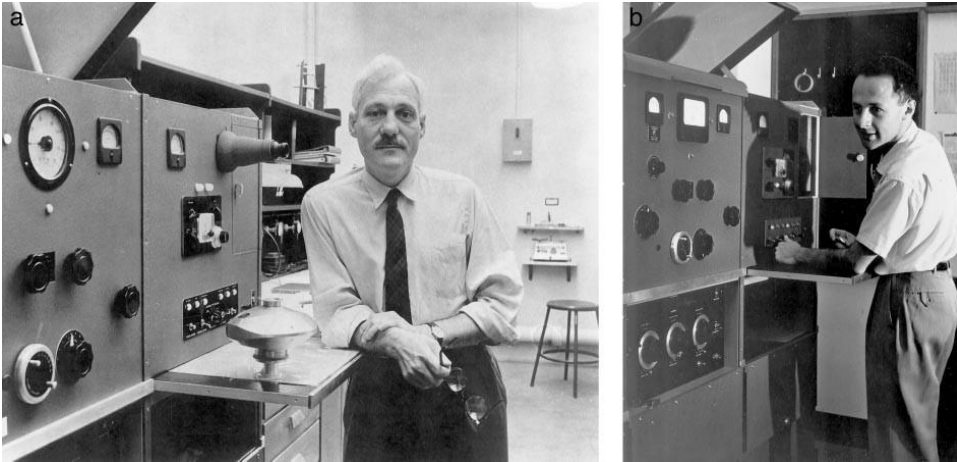
2:30 h

Centrifugation time

Gray, *Scient. Amer.* **184**(6):42-51, 1951

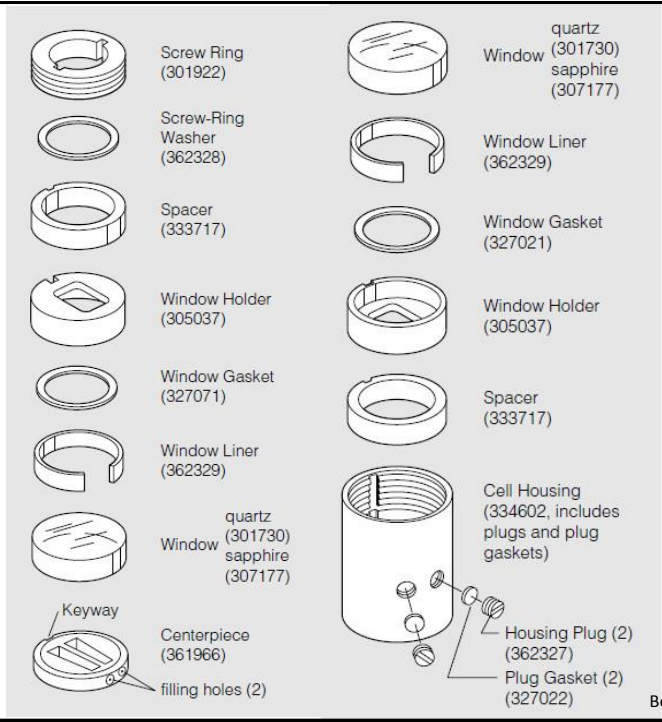
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Jerome Vinograd and Mathew Meselson and the analytical ultracentrifuge *Spinco Model E ser. #168*



Hanawalt, PNAS 101:17889-17894, 2004

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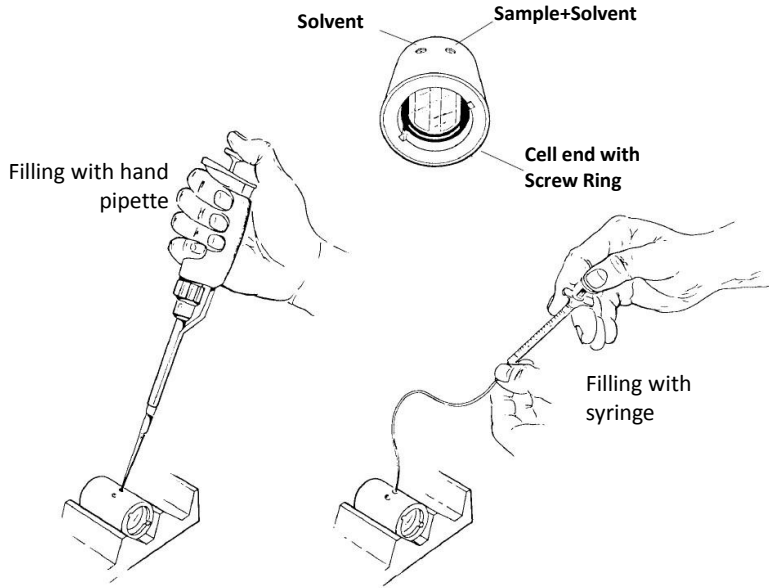


Mounting the cell of a modern analytical ultracentrifuge rotor

Beckman-Coulter™ LXL/A-TB-003F, 2001

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Filling the the two chambered standard cell

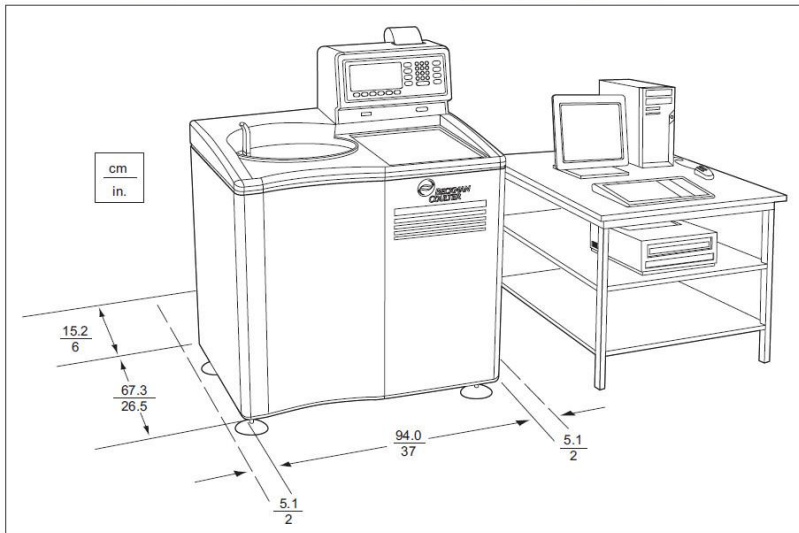


Beckman-Coulter™ LXL/A-TB-003F, 2001

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A modern analytical ultracentrifuge

ProteomeLab XL-A or XL-I Instrument and Data System Space Requirements



Beckman-Coulter™ PN LXLAI-IM-10AB 2014

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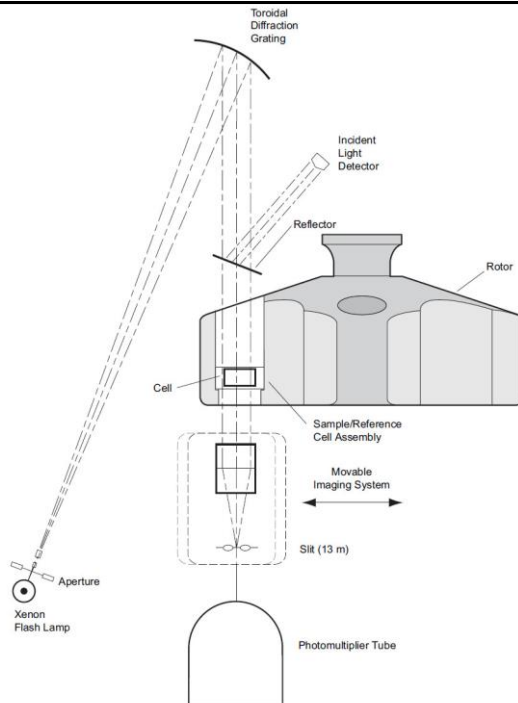
Analytical Ultracentrifuge Beckman-Coulter Model XL-1



Planken KL (2008) – PhD Thesis - ISBN 978-90-393-4798-0

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UV absorption measurement system In a modern analytical ultracentrifuge



Beckman-Coulter™ PN LXLAI-IM-10AB 2014

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HOW THEY DID IT

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I
EQUILIBRIUM SEDIMENTATION OF MACROMOLECULES IN DENSITY GRADIENTS
WITH APPLICATION TO THE STUDY OF DEOXYRIBONUCLEIC ACID

II
THE CRYSTAL STRUCTURE OF N,N'-DIMETHYL MALONAMIDE

Thesis by
Matthew Meselson

In Partial Fulfillment of the Requirements
For the Degree of
Doctor of Philosophy

California Institute of Technology
Pasadena, California

1957

**PhD thesis of
Matthew Meselson
1957**

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I am grateful to Dr. Raphael Pasternak, my research advisor until he left the Institute in 1956, for his good friendship and lively criticisms.

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The National Science Foundation and the California Institute of Technology have provided me with appreciated financial support.

In many ways my parents have made it possible for me to pursue this work.

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PNAS, 43:581-588, 1957

*EQUILIBRIUM SEDIMENTATION OF MACROMOLECULES IN
DENSITY GRADIENTS**

By MATTHEW MESELSON,† FRANKLIN W. STAHL,‡ AND JEROME VINOGRAD

GATES AND CRELLIN LABORATORIES OF CHEMISTRY§ AND NORMAN W. CHURCH LABORATORY OF
CHEMICAL BIOLOGY

CALIFORNIA INSTITUTE OF TECHNOLOGY, PASADENA, CALIFORNIA

Communicated by Linus Pauling, May 27, 1957

Meselson et al., PNAS 43:581-588, 1957

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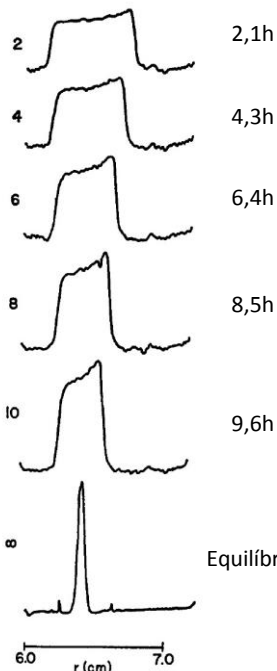
Resumo do método desenvolvido

The centrifugal field tends to drive the macromolecules into the region where the sum of the forces acting on a given molecule is zero. (The effective density of the macromolecular material is here defined as the density of the solution in this region.). This concentrating tendency is opposed by Brownian motion, with the result that at equilibrium the macromolecules are distributed with respect to concentration in a band of width inversely related to their molecular weight.

Meselson et al., *PNAS* 43:581-588, 1957

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Meselson e Nazarian (1963), in: J.W. Williams (ed.) *Ultracentrifugal Analysis*, pp.131-142



Ultracentrifugação de DNA de T7

In a CsCl solution at 20 °C with initial density of 1,720 g/cm³ a DNA initial concentration of 3 µg/mL

In a *Spinco Model E* Analytical Ultracentrifuge at 31.410 rpm

The time required for the concentration of a species to be at 1% of the equilibrium value of the center of a Gaussian with two standard deviations can be estimated as:

$$t^* = \frac{S^2}{D} \left(\ln \frac{L}{S} + 1,26 \right) \quad \begin{array}{l} L = \text{Length of the liquid column} \\ \sigma = \text{standard deviation at} \\ \text{equilibrium} \\ D = \text{diffusion constant} \end{array}$$

$L \gg \sigma$

Meselson et al., *PNAS* 43:581-588, 1957

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THE REBUTTALS

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Vol. 1, No. 3 BIOCHEMICAL AND BIOPHYSICAL RESEARCH COMMUNICATIONS Sept. 1959

THE SUBUNIT OF DEOXYRIBONUCLEIC ACID*

1919 - 2013

Liebe F. Cavalieri, Barbara Hatch Rosenberg and Joan F. Deutch

Sloan-Kettering Institute for Cancer Research,
Sloan-Kettering Division of Cornell University Medical College
New York, New York

Received September 8, 1959

The DNA molecules of *E. coli* B have been shown by Meselson and Stahl (1958) to be composed of two subunits, one of which is parental and the other, newly synthesized. Similar results have been obtained at the chromosomal level by Taylor, Woods and Hughes (1957). It is obviously essential to know the nature of the molecular subunit in considering possible mechanisms of replication at both levels.

Experiments carried out in this laboratory show that the subunit consists of two strands, rather than the single polynucleotide chain necessitated by the Watson-Crick (1953) replication hypothesis. Molecular-weight measurements by the light-scattering method confirm the results obtained by Meselson and Stahl, who used the method of equilibrium sedimentation in a density gradient, that the molecular weight of *E. coli* DNA is reduced to half by heating in cesium chloride. However, the DNA used by them has been found to be an aggregate held together by protein links; this type of linkage differs from that occurring between the two subunits.

Experiments carried out in this laboratory show that the subunit consists of two strands, rather than the single polynucleotide chain necessitated by the Watson-Crick (1953) replication hypothesis.

Cavalieri et al., *BBRC* 1:124-129, 1959

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DNA isolated from E. coli B grown under the identical conditions used by Meselson and Stahl and deproteinized by the Duponol method had a molecular weight of 11×10^6 (light scattering) and, when centrifuged in CsCl, formed a band identical with that obtained by Meselson and Stahl. The molecular weight dropped to 5.6×10^6 on heating to 100° in CsCl, and the band width increased. Heating in the absence of CsCl did not alter the molecular weight.

When the 11×10^6 DNA was either treated with chymotrypsin, or shaken with a chloroform-octanol mixture, its molecular weight dropped to $2.4 \pm 0.2 \times 10^6$ while its length actually increased. Repeated treatments

Cavaliere et al., *BBRC* 1:124-129, 1959

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- Knowing that a deoxynucleotide has a mean molecular mass of 324.5 Da, what is the size in base pairs of a molecule with 11×10^6 Da

$$\frac{11 \times 10^6 \text{ Da}}{2 \times 324.5 \text{ Da}} = \frac{11 \times 10^6 \text{ Da}}{649 \text{ Da}} = 1.7 \times 10^4 \text{ pb} = 170 \text{ kbp}$$

- Knowing that the genome of *E. coli* has 4.6×10^6 base pairs, how many fragments of 11×10^6 Da exist in the genome?

$$\frac{4.6 \times 10^6}{1.7 \times 10^4} = 271 \text{ fragments}$$

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The *E. coli* DNA is a dimer with two double helices!

We deduce that the unaggregated, unit DNA molecule of *E. coli* is actually a dimer composed of two double helices, laterally bonded together; that each double helix is conserved intact during cell division; and that the bonds holding the dimer together are ruptured by heating in CsCl, as well as by some part of the replication cycle in the cell. The dimer bonds are clearly

Cavalieri et al., *BBRC* 1:124-129, 1959